INDIVIDUAL WELL REPORT FROM THE PROGRAM ON

CHARACTERIZATION AND ANALYSIS OF DEVONIAN SHALES AS RELATED TO RELEASE OF GASEOUS HYDROCARBONS

WELL C-336 MARTIN COUNTY, KENTUCKY

bу

R. S. Kalyoncu, J. P. Boyer, and M. J. Snyder



BATTELLE Columbus Laboratories 505 King Avenue Columbus, Ohio 43201

Date Published - May 16, 1979

PREPARED FOR THE UNITED STATES DEPARTMENT OF ENERGY

Under Contract No. DE-AC21-76MC05205

INDIVIDUAL WELL REPORT FROM THE PROGROM ON

CHARACTERIZATION AND ANALYSIS OF DEVONIAN SHALES AS RELATED TO RELEASE OF GASEOUS HYDROCARBONS

WELL C-336 MARTIN COUNTY, KENTUCKY

from

BATTELLE Columbus Laboratories

DOE Contract No. DE-AC21-76MC05205

"This report was prepared as an account of work sponsored by the United States Government. Neither the United States nor the United States DOE, nor any of their employees, nor any of their contractors, subcontractors, or their employees, makes any warranty, express or implied, or assumes any legal liability of responsibility for the accuracy, completeness, or usefulness of any information, apparatus, product or process disclosed, or represents that its use would not infringe privately owned rights."

ice

LIST OF CONTRIBUTORS

- M. J. Snyder, Program Manager
- R. S. Kalyoncu, Principal Investigator
- J. P. Boyer, AID Analysis
- C. A. Alexander
- D. Chase
- R. W. Coutant
- J. W. Lennon
- D. W. Pickett
- M. Rausch
- A. J. Roese
- A. J. Walker
- M. V. Weaver

TABLE OF CONTENTS

				Page
INTRODUCTION AND SUMMARY		•	•	1
OBJECTIVE AND SCOPE				2
A. ANALYSIS AND DISCUSSION OF THE CHARACTERIZATION DATA		•	•	3
Task 1. Core Sampling		•	•	3
Task 2. Gas Contents and Release Rates		•	•	3
Task 3. Chemical Characterization Data		•	•	4
Task 4. Summary of Physical Characterization Data		•	•	4
Task 5. Shale Lithology		•	•	5
Task 6. Data Correlation Analysis	. .	•	•	5
LIST OF TABLES				
TABLE 1 WELL DATA FOR C-336	, •			9
TABLE 2. FIELD SAMPLING DATA FOR WELL C-336				11
TABLE 3. INITIAL GAS RELEASE DATA: WELL C-336				14
TABLE 4. STATISTICAL ANALYSES OF OFF GAS DATA		•	•	17
TABLE 5. CHEMICAL CHARACTERIZATION DATA		•		18
TABLE 6. SUMMARY OF PHYSICAL CHARACTERIZATION DATA		•	•	21
TABLE 7. STATISTICAL ANALYSES OF PHYSICAL CHARACTERIZATION I	CAC	'A	٠	24
TABLE 8. POROSITY DATA FROM DENSITIES		•	•	25
TABLE 9. MERCURY INTRUSION POROSITIES FOR WELL C-336		•		28
TABLE 10. ENERGY DISPERSIVE ANALYSIS OF SHALES				31

. .

LIST OF FIGURES

		<u></u>	Page
FIGURE	1.	ON-SURFACE TIME FOR WELL C-336	32
FIGURE	2.	HYDROCARBON GAS VERSUS DEPTH FOR WELL C-336	33
FIGURE	3.	CARBON VERSUS DEPTH FOR WELL C-336	34
FIGURE	4.	HYDROGEN VERSUS DEPTH FOR WELL C-336	35
FIGURE	5.	H/C ATOMIC RATIO VERSUS DEPTH FOR WELL C-336	36
FIGURE	6.	NATURAL GAS VERSUS CARBON FOR WELL C-336	37
FIGURE	7.	NATURAL GAS VERSUS HYDROGEN FOR WELL C-336	38
FIGURE	8.	HYDROCARBON GAS VERSUS BULK DENSITY FOR WELL C-336	39
FIGURE	9.	CARBON VERSUS BULK DENSITY FOR WELL C-336	40
FIGURE	10.	HYDROGEN VERSUS BULK DENSITY FOR WELL C-336	41
FIGURE	11.	POROSITY VERSUS DEPTH FOR WELL C-336	42
FIGURE	12.	POROSITY VERSUS NATURAL GAS FOR WELL C-336	43
FIGURE	13.	POROSITY VERSUS CARBON FOR WELL C-336	44
FIGURE	14.	POROSITY VERSUS HYDROGEN FOR WELL C-336	45
FIGURE	15.	SEM MICROGRAPH OF SAMPLE C-336-3842	46
FIGURE	16.	SEM MICROGRAPH OF SAMPLE C-336-3315	47
FIGURE	17.	SEM MICROGRAPH OF SAMPLE C-336-3186	48
FIGURE	18.	SEM MICROGRAPH OF SAMPLE C-336-3186	49
FIGURE	19.	SEM MICROGRAPH OF SAMPLE C-336-3186	50
FIGURE	20.	SEM MICROGRAPH OF SAMPLE C-336-2646	51
FIGURE	21.	SEM MICROGRAPH OF SAMPLE C-336-2524	52
FIGURE	22.	C-336 HYDROCARBON GAS VERSUS LABORATORY AND WELL LOG DATA	53
FIGURE	23.	C-336 HYDROCARBON GAS VERSUS WELL LOG DATA	54
FIGURE	24.	C-336 OPEN POROSITY VERSUS LABORATORY AND WELL LOG DATA	55
FIGURE	25.		56

INDIVIDUAL WELL REPORT FROM THE PROGRAM ON

كافية والأرا فالمتديد للكافيات

CHARACTERIZATION AND ANALYSIS OF DEVONIAN SHALES AS RELATED TO RELEASE OF GASEOUS HYDROCARBONS

WELL C-336 MARTIN COUNTY, KENTUCKY

by

R. S. Kalyoncu, J. P. Boyer, and M. J. Snyder

INTRODUCTION AND SUMMARY

Partial data on the characterization of C-336 (Martin County, Kentucky) shales were reported in the Sixth Quarterly Technical Progress Report, January through March 1978. Total characterization data and interpretation are presented in this Individual Well Report on shales from Well C-336 (Ky #3).

Coring of C-336 well was accomplished in November 1976. Samples were obtained at depths from 2479 to 3371 feet. 119 samples were collected for Battelle and 90 for other DOE contractors.

Shales from Well C-336 exhibit relatively high hydrocarbon gas contents. Oxygen to nitrogen ratios in the gas analysis continue to be very low, leading to various possibilities mentioned in previous reports.

Wide range in carbon contents characterizes the chemical composition of the organics in these shales. Positive relationships exist between the hydrocarbon gas contents and chemical data (carbon and hydrogen). Relationships between the physical data and gas contents are not as distinct, though, an inverse relationship does exist between the bulk density and hydrocarbon gas contents.

OBJECTIVE AND SCOPE

The objective of this program is to determine the relationships between shale characteristic, hydrocarbon gas content, and well location to provide a sound basis for defining the productive capacity of the Eastern Devonian Shale deposits and for guiding research, development, and demonstration projects to enhance the recovery of natural gas from the shale deposits. The program includes a number of elemental tasks as a part of the Resource Inventory and Shale Characterization subprojects of DOE's Eastern Gas Shales Project and is designed to provide a wide variety of support data for that project.

A large number of core samples of gas bearing Eastern Devonian Shale will have been examined by the end of the program. After the characterization data for individual wells have been compiled, attempts will be made to establish the interrelationships between the shale characteristics, the hydrocarbon gas content, and well locations from which the samples were obtained, employing the Automatic Interaction Detection (AID) Analysis.

The following tasks comprise the total efforts in this research program:

<u>Task</u>	Descriptive Title
1	Coring Sampling
2	Gas Content and Gas Release Kinetics
3	Chemical Characterization of Shale
4	Physical Characterization of Shale
5	Lithology of Shale
6	Data Interpretation and Correlation

كالمناء والما والمنادة المناثانيات

A. ANALYSIS AND DISCUSSION OF THE CHARACTERIZATION DATA

Detailed descriptions of the experimental procedures employed in the execution of the elemental tasks described were reported in the Appendix of the Fifth Quarterly Technical Progress Report (ORO-5205-5) submitted to U.S./DOE in January 1978.

Task 1. Core Sampling

The coring of the C-336 Well (Martin County, Kentucky; API Well Code No. 31020) was completed in November 1976. The coring was begun at 2479 feet and stopped at 3371 feet, requiring 18 barrels to complete. 119 samples were collected for Battelle and 90 samples for other DOE contractors. Additional pertinent well information and field sampling data are provided in Tables 1 and 2.

Task 2. Gas Contents and Release Rates

The initial gas release data for Well C-336 are summarized in Table 3. The data include total hydrocarbon gases, nitrogen, oxygen, and carbon dioxide, all of which accumulate in the free space inside sealed canisters. Upon sealing the canisters in the field, the canisters are transported to the laboratory and stored for at least three weeks prior to "free" gas analysis.

A significant feature of the gas analysis results is the very low oxygen to nitrogen ratios, as has been the case with samples from all the wells thus far studied. Speculative explanations for this observation were presented in the individual well report on C-l shales.* Further research in the laboratory would shed further light as to what processes are taking place which result in oxygen difficiency in the system.

^{*} R.S. Kalyoncu, et al., Individual Well Report from the Program on Characterization and Analysis of Devonian Shales as Related to Release of Gaseous Hydrocarbons, Well C-1, published March 22, 1979.

Statistical parameters on the gas analysis are presented in Table 4. Total hydrocarbons average 16.57 percent by volume of total gases, methane being the major hydrocarbon gas with an average volume percent of 10.23. Average volume of hydrocarbon gas is 0.38 cubic foot of gas per cubic foot of shale.

Natural gas contents of shales vary randomly with depth as illustrated in Figure 2, though it may be possible to observe two zones; one where the hydrocarbon gas contents somewhat decrease with depth (between 2424 and 2824 feet) and one where the hydrocarbon gases increase with increasing depth (2824 and 3424 feet).

Task 3. Chemical Characterization Data

Chemical analysis data (carbon, hydrogen, and nitrogen) are summarized in Table 5. The carbon values range from 0.5 to 7.6 percent with hydrogen values ranging between 0.4 and 1.0 percent by weight of total solids. Total carbon values are plotted as a function of depth in Figure 3. No distinct pattern exists between carbon contents and depth, though bands of increasing carbon contents with depth are noticeable at three distinct intervals of depth, namely, 2425 - 2724 feet, 2824 - 3034 feet, and 3124 - 3424 feet. Figure 4 illustrates the hydrogen content of the shales as a function of depth. Three distinct bands of constant hydrogen contents with depth are apparent at values 0.50, 0.7, and 0.8 percent hydrogen. H/C atomic ratios, as presented in Figure 5, exhibit a wider scatter than either carbon or hydrogen values. However, a pattern resembling more closely the carbon profile is apparent from Figure 5.

Figure 6 illustrates the relationship between natural gas and carbon contents. A steady increase in natural gas contents with increasing carbon content is quite apparent. Three to four bands of constant hydrogen content are seen with varying natural gas contents, as shown in Figure 7.

Task 4. Summary of Physical Characterization Data

Partial physical characterization data are summarized in Table 6. Among the data obtained are densities, porosity, surface area and permeability. The statistical parameters on the physical data presented in Table 6 are given in Table 7. Bulk density values range from 2.64 to 2.68 with a mean value of 2.66. Porosities (calculated from the density data) vary between 2.43 and 3.29 percent with a mean value of 2.86 percent. Surface areas range from 2.25 to 2.83 having a mean value of $2.54 \text{ m}^2/\text{g}$.

A slight increase in the natural gas contents with decreasing bulk density values is somewhat apparent from Figure 8. Such relationship is much more pronounced between the bulk density and carbon values as illustrated in Figure 9. Hydrogen contents also show an increase with decreasing bulk density values. This is shown in Figure 10.

Figure 11 illustrates the porosity profile (calculated from the density data) of C-336 shales with depth. Increase in porosity values with depth is apparent throughout several depth intervals from the Figure. Natural gas contents show slight decrease with increasing porosities for the C-336 shales. This relationship is shown in Figure 12. Similar relationship between the porosity and carbon contents is apparent from Figure 13. Hydrogen contents, illustrated in Figure 14, also fit the similar pattern of natural gas and carbon profiles with increasing porosities associated with decreasing hydrogen contents.

The apparent observations in Figure 12 through 14 can somewhat be termed anomolous, as, for example, one could possibly expect high porosity values to be associated with high hydrocarbon gas contents. On the other hand, low hydrocarbon gas contents with increasing porosity values could be explained by the possibility that the hydrocarbon gases might more easily escape the more porous shale structure prior to sealing the canisters. This point once again illustrates the importance of making additional gas flow measurements at the well site to complement the laboratory gas analysis data on the sealed shale samples.

The porosity values calculated from the density data are presented in Table 8. Table 9 summarizes the mercury intrusion porosity values for C-336 shales.

Task 5. Shale Lithology

Lithology of the shales was studied using SEM/EDAX and the petrographic microscope as in the past with other shale samples. Table 10 summarizes the elemental analyses of seven C-336 shale samples as determined by Energy Dispersive X-ray Analysis (EDAX).

Figures 15 through 21 are SEM micrographs of seven C-336 shale samples. Five of the seven samples were relatively high in illite content, the other two containing relatively very high or very low amounts. Analyses supported this microscopic observation to the extent that the K contents of the samples containing the illite in the extremes were also the highest and lowest. Microscopically, the quartz content of all but two of the samples was judged to be medium. Two of the samples appeared to have a very low quartz content, these same two having a high illite content as noted both by microscopic observation and a high K and relatively low Si content by analysis. The sample appearing to have the lowest illite content contained the highest amount of Si by analysis. It did not appear, microscopically, to have a high quartz content, but it was heavily stained, which may have interfered with the quartz identification. Pyrite content of the samples was variable, ranging from scarce to abundant. Microscopic observation showed an absence or very little occurrence of carbonate minerals. This was generally supported by analysis. One sample, however, contained a significant amount of Ca, which was not accounted for in the powder sample examined microscopically.

Task 6. Data Correlation Analysis

Analysis of the laboratory generated physical property, and gas data and the wire logwell data has been carried out using the Automatic Interaction Detection (A.I.D.) technique. Results similar to those of

درون و در کانده در سیال

previous studies were found in this work, as we have again seen some good correlations between physical data (density and porosity) and hydrocarbon gas contents.

Figure 22 shows how the hydrocarbon gas contents are split when the potential splitting variables include all of the laboratory data and well log data. The first split is based on the bulk density and the split is normal, that is, low density is associated with high gas. The second split, based on gamma ray, lends further support to the value of this tool to identify regions of high gas. Based on our previous observations, however, the predictive quality of the gamma radiation to gas content is tied to the maturity of the organic material in the shale. The fourth split, made on closed porosity, is interesting in that the split is opposite of what might be expected. That is, low closed porosity is associated with higher gas. The explanation for this is not clear, but may be tied to the fact that the total porosity of these specimens does not show a great deal of variation, but the ratio of open to closed porosity changes. Therefore the split on closed porosity may indicate an increase in open porosity.

Figure 23 summarizes the results of comparing the hydrocarbon gas contents to the well log data. The most important observation here is that there is a correlation between exposure time or on-surface time (the time between the core leaving the well and being sealed in canisters) and gas content. The specimens with the shorter exposure time display the most gas. This points out the greatest liability of these analyses, that is, care should be taken in treating these gas contents as absolute, as they have undoubtedly lost gas prior to sealing. Furthermore, since the gas values show a dependence on porosity, it is not reasonable to assume all of the specimens lose an equal portion of gas even when the exposure time is the same.

Figures 24 and 25 show the results of our examination of open porosity. These analyses are particularly encouraging in that they show a very good correlation between Schlumberger's resistivity data and porosity. The correlation is that low resistivity is associated with high porosity. The same type of correlation is seen with the low porosity group in Figure 25 where high conductivity is associated with high porosity.

In figure 24 there is an apparent anomoly when two groups are split on the basis of density with higher densities being associated with higher open porosity. These splits are not particularly productive in that they do not markedly increase overall predictability. The explanation for this behavior is not clear. The probable explanation, however, is that this is a mathematical anomoly arising from the manner of calculating porosity based on density measurements. It may also be that as the shale is densified it is the closed porosity that is reduced.

TABLE 1. MELL DAIA FOR C-356

CUUNIY, NY

KAHIIN

LUCATION:

AL I I fuut :		933 FEET	ī.	E t. T											1	1	:
CUOKD INA 1 ES ;		37.46 (DEGREES.HINUTES) 62.29 (DEGREES.HINUTES)	3.3	pe G De G	X E E E	Z Z	INCI	(S)		LAII	LAIITUDE LONGIIUDE						
CUMING BEGAN AT 2424 FEET AND STUPPED AT 1371 FEET, IT TOUK 18 BANKELS TO CUMPLETE.	=	5452	FE	_	A S	15	UPPE	a,	_	1188	+ + +	_•) 	TOUK	2	A A	KELS
SAMPLES WENE KETUKNED TO BATTELLE UN 117 6/76.	KE 1	UKNED 1	201	145	TELL	J.	₹	=	3	76.							
INERE WENE 119 SAMPLES COLLECTED FOR BATTELLE AND FOR DIHERS.	7.	SAMPLE) 3	703	-t C 1	EU	FO.F	Ä	116	ר ר	ANU	3	တ	90 SAMPLES COLLECTED	, C0	1160	160

NUN INFURMATION FROM WELL C-336

TABLE 1. (Continued) KUN INFORMATION FROM MELL CONTINUED C-336

(

HERE	1 HE \$ E K E	THE MERE	JHE NERE	1HE KEKE	A ME MERE	JHE Neke	THE NEKE	THE	JHE 'NEKE	JHE Nehe	IHE Kere
ZBIS FEET. THE SAPPLES	2873 FEET. THE THE SAMPLES MEKE	2931 FEET. THE SAMPLES	2989 FEET. THE SAPPLES	3039 PEET. THE SAMPLES	SIOPPED AT 3097 FEET. PER FOOT, THE SAMPLES	3155 FEET. THE SAMPLES	3215 FEEL. THE SAPPLES	3272 FEET. THE THE SAMPLES WERE	SALZ FEEL. THE SAPPLES	347. THE	3411 FEET. THE SAMPLES
STOPPED AT PEN FOOT.	STOPPED AT PER FOOT.	Blupped af Per Foot.	STOPPED AT PER FOUL.	SIOPPED AT PEN FOUT.	STOPPED AT PER FOOT.	STUPPED AT PER FOUL.	SIOPPED AT Per foot.	STUPPED AT PER FOUT.	SIUPPED AT PER FOOT.	Slupped at Per Foot.	STUPPED AT PER FOOT.
AND CUKING 24 MINUTES KS	AND CURING 92 minuies 45	AND CORING 70 Minutes RS	AND COKING 67 MINUTES KS	AND CONING 90 MINUTES 85	ANU COMING 38 minutes 88	AND CORING	AND COMING 98 minuies 85	AND CORING 52 MINUTES 48	AND COHING 64 MINDLES 48	CURING MINDIES	AND COMING 28 minutes 25
CORING BEGAN AT 2757 FEET AND COKING AVEHAGE CORING HAIE 15 8,29 MINDIES ON THE SURFACE IN 2,85 HOUKS	CURING BEGAN AT 2815 FEET AND CURING AVERAGE CURING HATE IS 6.92 MINUTES UN THE SURFACE IN 2.13 HOUMS	CURING BEGAN AT 2873 FEET AND CORING AVERAGE COMING RATE IS 7.70 MINUTES UN THE SURFACE IN 2.55 MOURS	CURING BEGAN AT 2951 FEET AND CONTING AVENAGE CONTING RATE IS 6.67 MINUTES ON THE SURFACE IN 2.55 HOURS	CUKING BEGAR AT 2989 FEET AND CONING AVENAGE COMING RATE IS 4.90 MINUTES ON THE SUNFACE IN 3.17 HOURS	CURING BEGAN AT 3039 FEET AND COKING AVERAGE COKING RAIG 15" 6.38 MINUTES ON THE SUNFACE IN 3.50 HOURS	COMING BEGAN AT 3097 FEET AND COMING AVERAGE COMING HATE IS 7.45 MINUTES ON THE SUNFACE IN 2.68 HOURS	COHING BEGAN AT 3155 FEET AND COMING AVEHAGE CORING MAIE IS 7.98 MINUTES ON THE SUMFACE IN 2.67 HOURS	CURTING BEGAN AT 3213 FEET AND CURING AVERAGE CORING RATE 15" 4.52" MINUTES UN THE SURFACE IN 1.66 HOURS	CORING BEGAN AT 3272 FEET AND COKING AVERAGE COKING RATE 18 9.64 MINDIES ON THE SUNFACE IN 2.75 HOURS	CURING BEGAN AT 3512 FEET AND CURING AVERAGE CURING RAIL 15 TO 52 MINDIES UN THE SURFACE IN 2.24 HOURS	COMING BEGAN AT 3370 FEET AND COMING AVERAGE COMING HATE 15 9.28 MINUTES ON THE SURFACE IN 2.17 HOURS
BEGAN ATE CONTING	BEGAN ATE CONTING	BEGAN AT E COMING SUNFACE	BEGAN ATE CONTING	BEGAN ATE COMENG SUNFACE	BEGAN AT	BEGAN AT E COKING (SUNFACE	BEGAN AT E CORING SUMFACE	BEGAN ATE CORING SUNFACE	BEGAN AT E CUKING S SUNFACE	BEGAN ATE CURING I	BEGAN AT E CUMING I SURFACE
CORING AVENAG ON THE	CUKINE AVERAG UN INE	CORING AVERAGE UN THE	CUMING AVERAGE ON THE	CUKING AVEHAG UN THE	CURING AVERAGE ON THE	CORING AVERAGE ON THE	CORING AVERAGI UN INE	CURING AVERAGI UN THE	CORING AVERAGO ON THE	CURING AVERAGI UN IHE	CORTNG AVERAGE ON THE
~	20	7	3	=	ય .	3	3	2.	91	2	3
KUN NO.	ON	2 2	S.	NC.	N.C.	2	¥0.	3.	2	0	S.
KCK	X C S	2	Z S	NO X	N.	*0%	RUN NO. 14	ברא צכי	KUN NO.	יחא אם.	KUN NU. 18

TABLE 2. FIELD SAMPLING DAIA FUR WELL CSSO

ar u	⊶પાস ਵ ળ∠	2 - 2 - 2 - 2 - 2 - 2 - 2 - 2 - 2 - 2 -			가 ⊃ ∼ ૫ M 중 이 국 국 국 국 국 :
BAHKEL. No		: ୩ ମଧ୍ୟ ମଧ୍ୟ ମଧ୍ୟ ମଧ୍ୟ ଖଟ ଜୁ) நிறி நிறி இது இது இது -	សំណុស់សំណុស់សំ ង១១១១ ១។	0~~~~
ON SURFACE TIME					นากกาน การการการ การการการ
согик	7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7	N 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	20 20 20 20 20 20 20 20 20 20 20 20 20 2	N	7 7 2 2 2 2
SAMPLE 10.	C356+224 C356+2244 C356+2444 C356+2454 C356+2465	C336-2494 C336-2494 C336-25144 C336-25144 C336-2524 C336-2545	C336-2565. C336-2565. C336-2575. C336-2586. C336-2580. C336-2607. C336-2617. C356-2617.	C 3 3 5 5 7 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	C436-2703, C436-2764, C436-2784, C436-2784, C436-2784,

	.2 Auda	. (Continued)		
	FIELD SAMPLING D	AIA FOR BELL	C 536	
SAPPLE 10.	CULUR	UN SURFACE TIME	BARKEL NU	SEG NO.
C336-2802.	7 2	2.67		57
336-281	NSN	•	~	46
30-2B2	74 F	э.	.c	~ 7
336-683	7	01.4	5.4	10 C
	7 7 7	•	3	.
356=285	Z	. =	3 20	51
36-286	7 2	24.5	20	şç
356-287	N4N5	2.62	39	.5s
336-	7 N	1.75	حِي	ąč.
36-2	ਤ 2	96.1	ټ	\$5
356-290	ਹ -	کام دران		30
7.7.4	₹	71.7	,	ָהָ יי
36-691	JYKZ/1	7	rc	ָהָ ע פּי
320-055	200	0 r4	ب ح	ŭ 4
1567-05	***	בים בים בים	^ <u>.</u>	9 -
42-95	1 7	05.5	2	פיק
336-2960	₹	2.75	91	6.9
36-2460	2 N	2.67	2.	79
336-2970	7 W	2.60	01	
-95	3	£ . 65	3 .	.
356-2	47 : Z	\$0°5	3	79
77.77	7 7	2°5		99
3 36 - 300 3	7 3	66.6		, , , , , , , , , , , , , , , , , , ,
3.56-301	١ .	34.4		
336-3629	. s	3.76		2/
350-3038	N.	3,92	11	7.5
	3 2	•	16	7.4
356-305	₹ 2		2	
306	V Z		<u>u</u> .	٩
3.56-30	Z	1 × 5	21	
356-30			u .	
356-50	74 : 2		u .	7 :
350-30	N Y	44. I	u c	3
30-310	4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4			
3.56-5117	1/4/1	C 7 - 1	7 -	7 4
1915-04 45-44	1 / 1 / 1 / 1 / 1			, 4
	17175		7 (7)	
336-5146	1/6/195	מח א	2	1 23
-315	5614/1	•	E1	1 23
43	17६धर	1.10	7	Ð
C330-3110.	1/595	1.45	7.1	gr.

í

REAN EXPOSURE TIME STANDARD DEVIALION OF EXPOSURE TIME

TABLE 2. (Continued)
FIELD SAMPLING DATA FÜR WELL C336

SEL NO.	2 m d d d	7	, , , , , , , , , , , , , , , , , , ,	\$ 7 0 - 2 7 7 0 0 	7 4 7 7 7 7
N N N N N N N N N N N N N N N N N N N	4 4 4 4 4	ដែលស្លាស់ ស្មា - ក្រុស្ស ស្មាល់ ស		977777	~ 0 0 0 0
UN SUMPACE TIME		1744-77 1744-77 1744-77 1744-77 1744-77 1744-77	น > ๑ น = รถเล ส ศ ส - ก ณ ก ก ก ก	2 7 4 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	2 m 4 m m
COLOK	5674/1 5674/1 5674/1 5674/1	1	5672/1 5672/1 563/1 563/1 563/1	2 Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z	
SAPPLE 10.			1336-3274. 1336-3294. 1336-3294. 1336-3304. 1356-3310.	C346-3415. C346-3425. C346-3435. C346-3445. C346-3445.	CARPACA. CARPARAA. CARPARAA. CARPARAA. CARPARAA.

TABLE 3. CHILLS GAS OFFICES DATA: AFLI CAM

47/50/90

C. 101			3 7 1 3 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	12 x 12 1	4. 611.4	30 V M 1716	1.16.111.1	413 6 T to 441	total	MINOGEN OXYGEN MARIN	OXYGEN	CARBOA	OF SHALF	2.02 2.02 2.03
		-	•		:									•
4.4)	* * * * * * * * * * * * * * * * * * *	· •	7	÷ • • •	¥ = *	7	1.7.1	00.0	41.14	47.74	# 6 H	1.3H). H	~
=		:	1.1	7.5.	て ズ・シ	4. 40	1.71	00.0	44.50	74.47	4.70	1.39	1.16	∼
	11114.	* 13	11.00	1 : " /	10.00	1.74	4.4	00.0	E	15.04	1.24	7.14	74.	~
_	1		14.5	; ·	1.50	٥,٠	4=.	0.00	\$€.√	77.06	20.06	₹.	رن ت •	*
F331 - 2464. 1125.		* 44 .	<i>i</i> .	11.11	1.0.1	7.	٠,	0.00	14.73		3.47	7.45	71.	ζ.
-2473. 7110.		. 7.0.		10.00	4.55	4 - 1	.40	ر :	4 1.11	27.01	, 74	3.64	٠,	٤
10.0	÷ ; ;	.11.	٠.	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	٥٧ • د	د. ۲ د د د	dr.	0.00	26.27	61.23	7.33	4.13	04.	7
		515.	.43	ベベ・トー	F. / * F.	5°.C	4.5	00.0	32.24	57.44	14.4	3.51	14.	I
1500	. H (2)	5.45	-:-	7. ·	2.4.0	17.	00.0	0.03	xe.c.	11.72	15.51	3.04	١٠.	Ť
644-2414.7 740.		471.	٨.7	00.0	00.0	00.0	0.00	0.00	00.0	77.07	21.94		00.0	0.1
F330-2534. 175.			4.54	a. 7.	1.60	÷.	₹.	00.0	K. 15	74.60	4.50	1.46	60.	=
6332534. (110.	-	13.	75.	<u>=</u> ,	£ <.	7.4.	- ·	00.0	٠. ۵٠	74.30	14.57	45.	= -	12
		444.	2.25		1.24	Ŧ.	0.00	00.0	7. nh	CO. OT	15.53	1.45	٠°.	13
6434-2455. " 126.	Ī	4.14.		Ξ.	00.0	. 45	00.0	00.0	÷.	77.66	21.13	14.	٠٥٠.	1
	. 60.3	467.	# * * * * * * * * * * * * * * * * * * *	11.63	\ = . w	1.56	.76	00.0	71.17	70.07	7.64	٦,30		۲_
•	1011	417.	٠ <u>. د</u>	7 7 . T		64.	₹.	00.0	4.24	74.18	17.23	64.	97.	Ċ
6334-7574. 114.	1965.	366	7.7.0	÷ .	75.1] I	Ĭ.,	00.0	2 1 • 4	17.06	14.32	15.	٠١٠	7
	* 1733. *.	·1.1.	7	<u>-</u>	Hc.	٠.·	٥.	00.0	H5.1	79.21	14.67	14.26	4· ·	۱
		.150	12.0	00.0	0.00	¥ 7 .	9¢.5	00.0	14.	74.77	51.49	.17	₹.	<u>-</u>
raak-2664; 764;			1.74	<u>r</u>	0.00		er.	00.0	. 34	19.00	14.29	1. 3н	00.	20
		*• In •		4.17	1::1	~	F. 43.	00.0	H . O.H	77.20	13.20	= 4 • 1	ac.	7
633-2617.7 750.	11.55	752	4.14	0.0	00.0	2	0.0	00.0	₹.	17.24	22.36	2 .	₹.	53
6335-3617. 170.		. t ac	7:	₹.	<u>:</u>	e e	0.00	00.0	e *	74.34	21.21	7.5.	=	₹
16.11		624	, T. s.	*	00.0	٤.	0.00	0.04	7	14.77	21.76	34	ē.	£ ,
		・だくご	/	- + · c -	1.51	٠, ٢	/ c.	00.0	F 6 * 6 T	67.56	2.79	ر د د د د د	7.	£
		o 67.	1.17		45. * 5	7.67	J • 4 C	~ =	64.13	n3.07	7.66	4.55	• 34	7
6 3 30 - 366 54 7 110.		40.4	/ 4.	0/*	4.06	4.62	r r	፤ .	٠ <u>۲.</u>	79.78	1.32	٠,٢	₹.	₹
		E & &	446	1.65	ر. چ	۲۰,۰1	18.	0.00	17.63	70.19	7.32	3.57	~c.	N::
	·	• c. (++	£ .	7. K	75.7	۲.42	٠.	<u>.</u>	15.71	68.03	11.40	5 ; ~ ;	٠. د	= ;
<u>.</u>		÷ ; ;	<u>.</u>	₹ :	٠	77.	77	e :	- K	/ 3 . 1 .	5. v	٠. د د د		= :
Control of the contro	-	* * * *		/ * *	; ;	, s - c	٠ - (9.0	= u		5 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6	\o.\ \o.\	``	\
		• • •	٠ •	7		•		• 3	C	0.1	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \		\	
	- 0 4 2	•	e :	7	7.		\ . :	90.0			0	9 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	* *	: :
				•	<u>.</u>		•						•	2 ;
		• \ \ - \ \ \ - \ \ - \ \ - \ \ - \ \ - \ \ \ - \ \ - \ \ \ - \ \ \ - \ \ \ \ - \ \ \ - \ \ \ \ - \ \ \ \ \ \ - \	_ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	•	= : :	T (÷ .	e :	= 0		18.76	۲ ٠٠ ۲	٠ <u>٠</u> ٠	Ę ?
•	-	•	•		* :		<u>-</u> ;			T. C.	•		·	= :
	:	اد	e :	\ - -			•	1.	\	17.13	T (£ :	- :	<u> </u>
<u> </u>		• • • • • • • • • • • • • • • • • • • •	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	1 T	<u>`</u>	<u>₹</u> ;	€.	0 · · ·		1.13	10.11	7.	<u> </u>	<u>.</u>
•		• • •		£ .	\		<u>c</u> .	3 · c	\	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	1.	* * .	E .	-
(44 47. 4 750		* 4:0	₹ : *. •		-	₹ : •	٠ د د	٢.	4.00	H	4.	r :	<u>:</u> ;	- ′
•	_		= -	- - -	₹ -	7.	,	-		75.09	٧ - ١	T	£ :	4
-7144.1 (25.		7. 7.	~ .	5 t. • 1	7.	÷.	• 0.1	٢.	٠. ځې	71.35	14.41	- H.	<u> </u>	*
1.71	•		£.,	•	<u>'</u>	÷.	* u *	0.00	- -	74.36	14.55	Į,	- °	J .
16.5	_	•••	= .	<u>.</u>	7.	, ,	٠ •	ee.e	15.	7.61	٠٠. ١٠٠4	<u>₹</u>	= .	C T
: 4/											•			

TABLE 3. (Continued)

1 dillate and sell-rafe inflate with Cline

04/55/10

SE 200	<u>-</u> 4	4	4	S.	۲.	55	53	.t.	55	ic i		i J	6.0	-	45	Ē	44	£ ;	00 7	- - - - -	69	2	7	22	:	٤ ٤	: £	11	æ ;	5.	E :	- n	Î	434	ţ	46	h 7	Ŧ	7 ·	9
OF SHALF	e e		20.	50.	.12	.22	7 = •	. 65	~~·	¥.;	***	- -	10.	æ.	¥1).	4%*-	~s	1.46			4°.	50.	Ξ.	07.	C 7 .	7 3	er : r	71.	97.	1.46	ر <mark>- د</mark> ۲	r (. `.	47.	Ξ.	Ξ.	.1.	٠ 9 9	₹°	
CARHONA DIOXIDE	2,82	I,	5.4.	1.45	1.40	6.05	4.62	6.53	1.71	4-24	/ H - '	70	7.05	6.13	1.41	1.21	1.52	- 4 - 4 - 6	<u>.</u>	79.	7.4%	1.17	.12	1.57	25°	7.	7 P - 1	1. 14	1 . n.4		3 ° .	5 C C		3.29	3.07	12.51	6.07	£9.	3.67	
	17.4	21. He	71.67	14.60	13.40	۶. د. ا	11.55	3. F.	26°51	ハマ・エ		50.4	10.90	.74	11.62	1.21	5.26	# :	7 J	15.91	3,37	17.96	14.74	14.48	10.65 2.	- ~ • d		4.94	05.	. E.	. 4°.	66.00	 	11,74	15.61	1.17	04.1	16.15	11,37	
NITPOGEU OXYGEN	14.47	75,49	77.16	77.48	74.10	65.69	74.76	n4.50	17.71	76.34	ر الم د الم د الم	10.50	73.59	44, 79	75.75	44.41	52.42	74°45	14.1	86.97 86.98	70.15	12.97	77.01	76.78	~	30 · 30 · 30 · 30 · 30 · 30 · 30 · 30 ·	41.47	74.61	27.20	٠. ٢٠	Λ.	K +	24.67	00.41	14.42	19.43		74. HT	47.15	
HALLANE PEGTANE HYDERG, WIT	5 5 1	1.17	₹. -	1.41	70°4	011.77	4.90	21.00	4.05	10.41	بار د د د د د د		***·/-	43.33	н. 74	45.17	40.25	11.00	7 7 ° 7 ° 7 ° 7 ° 7 ° 7 ° 7 ° 7 ° 7 ° 7		50.15	7. 14	7.07	いまって	41.17		17. 96 19. 96	19.00	10.71	47.35	u/. +0	7 m		10.04	5 Y . X	7.5.4	7.32	1.7H	1.70	
PFUTANE	00.0	00.0	00.0	00.0	511.	7.4.	0.00	00.0	80. 0	7	e :		÷		. 13	77.	<u>.</u>	60.1				74.	.45	00.0	æ. :		3 7	34.	£9.	1.40	00.0	00.0		00.0	00.0	10.	, I 4	00.0	00.0	
40 V 100+	2	5.	÷	÷.	٠.	. 72	7.	7.	<u>:</u>	44	₹.			1. An	. 33	1.64	2.	1.7. 2.5		2.6		<u>-</u>	36	ึกก	7 C		7.7	* U * T	7.11	- 50	٠. د .	\	=======================================	· •		٠,	1 4.	₹°.	/ u·	
Algorathera	c 7 .	7	64.	64.		1.11	1.12	H5*c		[. T			: 	5.0¢	7.4.	4.24	15.47	44.4	÷ ;			=	1.17	. 64.	₹. ~					15.3	e	[] () () () () () () () () () (ξ	7.15	{	1.1	An 4	なべ・	***	
40.91	Ţ		=	÷.	H., -	4.14	~·	7.5	•	7	÷ ;		· .	3.5	÷.		7.11	٠. د . د		· .		\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	٠٠,	٠ ۲	7.4	C4.41	?		12.05	55°/	₹. ₹.	֓֞֞֞֞֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓		~ .	·.	٠.	4. €.	57.	dr.	
- F-10-14	ž	ं ज 	44.	ì		4.5.5		57.8	37.	7.54.7	;; ·		77	47.67	4.70	10.00	****	<u>.</u>		داء. دع. ه	22.0	1. H.7	- u		₹	//*/* //*/*	2	11.11	- c	11.14			- 7	* * ° ¢	: · ·	4.5.3	1.24	70.	7,4 .,	
entustty ect.	81.0			2.05	- A	4.1.4	2.34	÷ - : +	2.4.	. 76	t	- 3	1.47	37.	4.41	£.7.	`.	£ :	37.9	 	4.71	. 37	£	T = 1.	ر د د د	\ .		4.13	. 715.1	٠ ٢ ٨	·.	0 4 1		1	4.44	* t	37.3	÷.	4.12	
1 1 1 1 N	947	5,77		204	475.	4/10.	44.3.	1,4,1	٠٠٠٠٠	441	.07		47.1	45.40	151	47.1.	150	= '.	<u>.</u>	. 7 e. 7 . 7 e. 7 . 7 e. 7	76.7	7.80	100%		. 7			0 44	F.H4.	42.7.	•		• · ·	1.1.1	267.	٠, ١٠,	• , (110)	134.	666	MI AM AR
3 10 1 0x	9 7 9 1	100		197		.01.6	947.	1.45.	110%	1042.	* 7 ! !			71.11.		.,,,,	1.1.	- - -	7.7	*	1015	11/11			.11.	• ,		• 1) - 1, 1	100.	* ***			• • •	1 1 1 1			H455.		7.44.	
Pur Sound Takes		. 0.97								750.	1.1.		2	1001	-2.2	. 0.0.4			· 4.5			·	7.50	1.71	. 1.5.1		· · · · · · · · · · · · · · · · · · ·	•00		• 3661	• = 5,			; ; ; ;	::	111.	125.	10.01		
ul Ahavs	000000000000000000000000000000000000000	33. 7. 7. 7. 7.				. 33 246.2.	C16- 41J	C 13 2.14.]	CHASSAUL.			C 4 4 7 7 4 1 1 .	11.00-011	1134 - 344 H	2 4 3 K 24 - 11 .	1330-2469.	C33.,-2969.7	9711-18	C 3 45 - 24440 .		**	C33<-3014.	C430-4019.7	. 136.23624;	6334-3034.	C 4 45 - 1040.	C 4 45 L 1855 4 .	C330-1059.7	C 3 3 30 7 4.	しょうし しゅいり	C 3 30 - 10 0 0 0 .	C 44% - 414 /	72.13.1.33	(33317)	1.7111-417	C 33%- 1196.	C 173-1147.	C330-3160.	C 133-1116.	

TABLE 3. (Continued)
[+17] [a], 558 SEL 619

04/06/79

	1811 43 100	WOM IS 4	Vin Hea	Vin 169 1991 OK 118			\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \			() of 05	=		Сакном	SAS FELFASED	860
al Thores	2.81		: : (- : (-	-1.34	4-1916	and the John	400 good	मात्र प्राप्त	to vit vot	нүйнөй.	N Tangfin	OXYGEN	OTOXIDE	OF SHALF	. ON
A. 44-(4-3)		:	. 88 c	4.6.1	74.	= \.	4.	έ.	0.00	£ 2.	74.35	18.94	14.	&¢••	2.5
C 13' - 31'56."	1	10 2 4	464	4.5.1	7 3 7	* 4 *	~ · ·	ne:	0.0	50.7	74.64	17.14	1.23	4.	63
C335-4205.		14.6 A.	. 1. G. C.	Tu . 5	14.7	24.	\$4.	٠.	[e ·	1.71	47.54	13.52	د .	. 0 •	40
6330-3216.	-= T T	÷ ;	<u>:</u>	e. 7e	·	. 4 ℃	7 4.	÷.	7.	15.5	76.05	70.41	36	<u> </u>	45
C334-3224.	1200		. 445.0	7.4.7	11.07	75.4	14.4	1.13	Ŧ.	41.05	54.66	٠٦٠.	4.54	45.	46
C 336-1244.	75.0	.474	515	4.5.4	1.73	÷.	44.	٥.	00.0	2.60	40.09	15.64	19.1	4 ii •	16
C 333-44.	720.	*1 -1-	4 17.	1.,.1	T : -	Ξ.	* 1 .	۷.	0.00	÷	00.67	14.54	3.43	50.	H
C 17 - 1764 7	146.	1446	. 4mf	4.6.1	. 17	.13	٠,	ī.	00.0	11.	7H. 74	19.06	4H.	ر».	ታ
6 135- 1254.	44.0	*	4.4.6	1.00	= = = = = = = = = = = = = = = = = = =	1.74	< E.	÷ 7.	√= .	13.65	14.21	10.64	1.47	۲٠.	001
C330-3754.	. 1411	.071	4 Mile.	2.45	04.1	47.5	H6. <	. 7.	.17	74.17	44.62	7.	h.27	. 7.0	<u> </u>
C 13n- 1274.	140	. 	. 6.4.	**	4.41	1 . He.	1.63	<u>.</u>	ro.	11.36	74.27	7.33	50°C	Ξ.	102
C336-1244.	7.40	, 11s.		4.6.4	10.6	· · ·	1.19	7.	٠, ١	44.0	40.91	5.A6	h.75	40.	103
C33 4994.	140	., ,,	, ; ; ;	19.4	30.5	0.00	00.0	00.0	00.0	00.0	80.70	1н.73	15.	00.0	104
C333356.	r.f.h.	1.1	5.4.5	4.44		£ 4.	< 7.	•	00.0	5.79	84.05	6.02	٠١. 14	. 03	105
C 178-1104.1	7.46	1053.	. 7.5.	1. 5.1	<mark>ا.</mark>	۲.		٧٥.	00.0	1.45	H4.26	6.56	2.0	40.	106
C 13h-1315.	1.5.	400	547.	f. 4.	* ! . .	2.73	۲۰۰۱	14.	. 1 S	11.39	74.19	6.75	5.1e	. 22	101
C336-3335.	.44.4	477.	473.	- X • I	£1.0%	ر. د	F. C.H.	1. CH	11.	70.04	66.39	7.55	1.45	07.	アニー
6336-3466	• = 1 •	*HY	544.	77.7	10.04	1.34	7.45	ζ.	٠٦,	17.44	74.03	1.14	7.37	6¢.	U [
C 37 - 37645.	1.11	, r = = =	4.17.	11.4	44.44	1.75	₹ -	1,0.	e.	₹. ₹.	14.70	9.12	2.27	<u>ه</u> .	==
C336-33457	74,11		44.7.	₹.√	7. +	2.44	1.97	7.7.		74.71	70.95	ع. د ک	00.5	£1.	<u> </u>
C338-3345.	1974	· \ - :	56.24	4.44	70.5H	٠,٠	4.44	1.14	£.	40.04	61.87	١.١۶	4.33	£y.	=
C33c-1364.	. 202	3.7M.	575	= ;	24.03	H. 75	٠, د ۲	۲۰۰۱	44.	41.73	64.64		4.93	14.	114
C 33 3475.	44.0	· :	6,2.1.	, f. /	10.03	۲ו ۱ ا	7.57	1.46	4.	14.14	14,03	0č.	3.56	. A 3	دا ا
6336-3344.	- 47.5-	403	547.	1.55	5, F . F	4.64	2.10	x.	.22	44.54	1.41	91.11	<u>.</u>	15.	911
1 33 3 1.14.			· 1.10	7.5. -	10.41	10.02	5.4.4	1.4.1	Œ.	H# 1.7	49.52	.63	2.63	41.	113
F33.1-34.14.	13.5	n67.	7.11	- x	0.4.16	D. 0H	28.4	.71	₹.	41.30	62.51	1.63	4.45	. 30	<u>- ۲</u>
F336-3404.1	146.	. 1212 н	556.		٠١٠.	. 65	1 .	00.0	00.0	5 7 .	79.32	20.05	42.	7 0.	<u></u>

TABLE 4. STATISTICAL ANALYSES OF OFF 64S HATA

61/20/50

WFLL 6336

	MEAN	STANDARD DFVIATION	VAHIANCE	COEFFICIENT OF DEVIATION	95 PCT. CONFIC LOWER LIMIT	95 PCT. CONFIDENCE INTERVAL LOWER LIMIT UPPER LIMIT	NO OF SAMPLES
MFTHANE, PERCENT	10,233	11.13	123.77	1.09	4.019	12.447	100
FIHAME. PERCENT	3.477	3.12	13.47	1.07	2,736	4.218	100
PROPANE. PERCENT	2,085	2.27	5.17	60.1	1,633	2.53A	001
AUTANF. PERCENT	665.	. 17	65.	1.30	.439	.744	100
PENTANE. PERCENT	.184	24.	.17	2.27	.101	.266	100
TOTAL HYDHOGAHRONS, PCT	16.571	17.47	305.11	1.05	13,095	20.047	100
NITROGEN, PERCENT	665-69	12.66	160.27	90	70.79	72,113	00
OXYGEN, PERCENT	11.144	7.03	49.40	.63	9.745	12,543	100
CARRON DIOXIDE, PENCENT	2.147	2.37	5.61	9#•	2.276	3,218	001
GAS VOI UMEZSHALF VOLUMF	.376	.42	.1.	1.10	. 293	.460	17

*! IS INSIGNIFICANT

v

		230.40.4	N Jeway IN	200 C 100	2	CODE	0
					#		
6336-2444	7.	0 • 1	-			\$	C.
336-2456.	7.2	♂ •	. 2.			S	·~
C 3 3 6 - 24 54 . /	6.7	1.0				5	,
6.336-2464.	- -	. ·	ν, σ			، ء	<u>د</u> م
336-2496			y			e ur	۰, د
336-2584	· ·	: ·	• -			٠ ٠	- 1
136-2514		9					; • •
3 16-2514.7	; • • • •	~	~			1 41	7 0
136-2524.	1.5	٠,	.2				
C 3 36-25 34.	- 1	٠٠٠	-			ۍ.	12
3.36-2445.		1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	:	***************************************		6	13
C 136-2555.	1.9		-			5	14
336-2566.	6.5	111111111111111111111111111111111111111	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	***************************************			15
F336-29665.7	1.6	•	:	•		S	91
r 136-2676.	3.	٠.					17
C * 36-2686.	٠.	y.				9	8
336-258A.	y.	4,0		***************************************	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		19
3.16-2546	1,					5	2.0
C336-26e7.	.	٠.	-		: 4		2.1
336-2617.	1.4	• 6	-1.			2	22
6136-2617.7	5	ų.				5	23
C336-2637.					1	\$	54
336-264F.	٠ مي	•	٠.			9	52
3.56-2656	5	9	•			٠ - ا	26
CSSh-Zbbbb.	~ .	، ت <u>ن</u> •	~.			in i	27
5 (E-2) FEF.	9 • 6	•				2	
6336-2676.	w.		~.			5	5-6
126-26H6.A	~ .	~			1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	ç	30
3.56-25.85.4	/ · · · ·	•	•••			r.	31
\$ \$6-7ng(•	3.1	, , , , , , , , , , , , , , , , , , ,			1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	S	32
C 5 55 - 77 H5.	e :	•	-			s i	· ;
216-2206	C • 4		1.			٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠٠	2 1
C. 3 3 0 - 7 C C 5 3 .	u •	•	u •			r,	S ?
27.70.70.00	7 .		•				2 5
446-2745.	, c	÷ •	• •				د . م
126-2/66	- C				1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		0.8
6.5 * 6 = 2 7 6 3 .	- 0	• •	÷ -			. · · • •	ית מית
336-2773.	3.2		-				- -
336-2784.	: O • s	, , ,	. 2				7 7
336-2784.7	-					د د د د	, ,- - J
1076-519	• •		- ·			, 4	7 -
C376-2482	0 1		•			د د	J .
		o v	-			r	2

TABLE 5. CHENICAL CHANACTERIZATION DATA, 3778

		COLAL, PERCEN	1	MPERATURE ASH,	CARD
VAMPLE I'I	CARREI	HYDPOGFR	NI TROSEN	200 C 100 C	3000
46-2430.	• • • • • • • • • • • • • • • • • • • •	4			
F= 46-2442.	2.0	7.	.2		
-2442.7	•	ن •	-		.
2462.	1.7		.1		
-2462	2.4	و التي			5
5-2472.	2.5		2	《曹书》《李生学》《夏书学》《中华》《中华》《李子文学》《李生学》《夏丰学》《夏丰学》《李子学》《李子学》《李子学》《李子学》《李子学》》《李子学》》《李子学》》	3
-2491	1.9) (5
6-2891.	٠,	ن •			S
6-2101.	2.1	· ·	. 2		5
6-2481.7	ς.	.	.1		5
6-2411.	3.5	2	2,		9
6-2121.	¥.£	•	~;		3
¢336-2131.	4	٠, ي	-5		5
6-2446.	6.5	٠.	~.		5
-2450.		9	.1		می
296.P.	7.3	6.	٠,		S
1-2460 L	F 9	Φ•	2		5
5-2470.	٠. ع.	· ·	.2		S
h - 2') A ()	9.	ی	•		S
C336-2498.	F. 5	æ.	•5		5
- 544d.					S
- 200a.	,,,	្រុ	. 2		יט
1-3019.	¥°.	2			5
5-3019.7	9 • : :	.	1.		\$
, - 31129.	9	7			3
5-3134.	6.6	6.			5
-3169.	¥. C	6, •	.2		5
3464 •	£ . %		~.		2
6-3069	50 m	C	~		S
6336-3069.7	8°2	٠.			5
6-3079	4.4	7	-		5
F336-3489.	5.3	1.6			
6.336-3990.	7.1	ъ.	.2		2
6376-7107.	5.0	7.	2,	Service and the service and th	5
	,		•		(

IJA IA
-
<u> ح</u>
2
Ξ
Ξ
A 1 100
~
7
CHARAL FLH 12
×
فند
-
7
2
Ï
=
Ξ
_
4
PHYSICAL
-
_
37
7
3
_
_
Ξ
~
ž
Stiffien Aid X
-
Ξ
ī.
ġ.
TABLE
=
<
-

		- 1	PC1	PC1	11/211	2 75
556-2454.	2,505	ςης.	000.0	4 04.	0 r r .	-
156-2444.	2.462	نولاد.	₹.	1.11.1	•	ر ا
\$56-2454.	2.465	/70.	1.731	2.5/B	5.440	-
56-2454.2	477	C. 44.	0.000	7.pp.7	3. C. 3.	3 1
56-2474.	5/5-2	062	300.0	0.57	27.5	- o
156-2474	2.515	-	405.	041.1	5. H.C.	•
\$6-2504.	2.554	. 3	000.0	044.	T	- 33
16-2514.	2.115	1.121	•	1.121	•	5
1.4162-05	2.585	10/	· •	491.0	. •	97
50-224.	2.640	6.304	2.5.5	4.55	3,500	=
56-2554.	6.11	7.7.1	000.0	7.7.	2.084	2
\$6-2245	2.754	4.195	0000	C41.5	042-4	5 1
\$6-255.	2.891	c48.1	000.0	•	= 10. =	7
\$6-2565.	699.2	3	2.017	9	3/5	<u>-</u>
7.5952-9	2.127	1,024	3.561	ς ρ 1 κ ς	215.1	4
\$6-25/5.	407°2	717.7	0.00	717.7	204.	2
25g	2.728	1,080	0.00	c	4.R2U	\$
\$6-258A.	2.706		330°3	4.411	2 × ×	3
36-245.	2.730	1,756	000.0	`.	37. ·	อา
30-C0U/ .	07/7	602.2	/14-1	3.060	000-1	v :
50-5017. 56-2017.	7117	7 4 4 7 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1			377	\ \ \ \ \
100	2.718	7.7	000.0	۱ ،•		12
Sn-264n.	669.2	4,1/1	000.0	4.171	94.5	.
\$6-2650.	2.650	244		150.7	1000	ુર
6-2066.	2,645	7	2.203	_	270.1	23
199-2000-1	2.603	7	0.000	17.1	nay.	#?
in-26/0.	2.650	7	5.119	961.4	5.450	67
497-	2.571	. 550	•	1.506	1,420	<u>جو</u>
) - -	166.7	154.	5,420	1000		\$\$
	040.7	cco.	30.0	כני.	2 7 7 ° 4	# ·
50 - C 1 C 2 C C C C C C C C C C C C C C C C	2 4 7	010	0000	010.		2.3
275	9		415.4			Ţ
\$6-2765.	2.6/4	764	0.00.0	•	002.8	3
0-2175.	9	144.	9.00.0	. 54.	3,690	77
36-2184.1	3	. 152	00000	. 132	270.7	4.5
\$6-2014.	4.714	, 854	0.000	, 654	010.1	46
156-2450.	7.117	1.514	0.00	1.512	1.000	44
56-284c.	80.	アオコ・一	000.0	K 10 0 1	4.130	50
	2.103	4.065	٠.	590.2	7.8.0	50
70''	9.4.2	=	1.2/8	4.52.5	4.150	75
6-2081.	9	777	<	1		34

TABLE 6. (CONTINUED) SURBBARY III PHYSICAL CHARACLERIZATION DAIA

	22/9	104	PC1	HC.1	MZ/1,	SEG
						•
C\$36-2891.		580.1	0,0,0	1.683	1.016	ቲቲ
C \$ 56-2901.	2.107	.162	0.000	.164	3.360	ąc,
\$6	671.5	050.	2.546	4.176	4,040	57
5-955	2.62h	. 514	000.0	+1¢.	1.000	X 1
\$ 50	444	37.	000.0	0.52	••	T
5.56-245	•	•	000	7/7-1	3	9.
\$.	•1	7	# / V * V	244.5	30.1	5
156-246	2.	70.	C00.	6/9·	3.00	.
0.2.0	• (200	000.0	200	000.0	3 4
162-056	950.2	0000		0000	050.1	9
000	2 3 3	U 4		3.3.5 3.5 3	07.7	
9		707.		664.	001.0	3
250-64		9 7	718.1			200
2 101 - 053	•	777		;	DAIA INCUMPLETE	? :
200	00u•2	700		*CC *	364.	-
1000	٠	W10*1	300° 4		•	27
556-5056	910°2		באר ה מומי ל		2 2 2	9 3
7707	•	7 7 7	999	- 4	3 7 C	
C 550-3034.	n	190.	200	# S S S S S S S S S S S S S S S S S S S	050.1	2 5
136-3007		"	3.020	3.73	0 2 7 7	
. 45	•• •	187	2.647	2.530	06/	67
556-509		365	000.0	C 9 C	020 t	300
156-	2.500	450.	070.0		0 7 7	19
	•	6,045	0.000	2.645	1.650	20
156-5121.		140.5	900.0	160.5	4.600	88
3	2.611	4.027	1 7 4 1	4.568	2.540	44
C 556-5137.2	99	4.451	00000	4.431	3.020	85
15-455	•	5,954	000.0	3.950	٠	a P
36-51	•66	156.8	5.718	11 P 7 F 6	5.000	<i>1</i> 8
18-955	9	169.5	• •	160.5	1.450	NO.
\$ \$6-31	٠	5.1/1	1.549	4.720	5.360	59
se- 31	9	スカコ・セ	975.	7,004	0/0.0	3
320-31	.67	5.078	0.000	5.070	5,430	ሪታ
\$ 5e-	• •	162.5	000.0	167.4	dou.5	2.2
C 556-5206.	•	0.00¢	970.0	ა ი ი ი	0.00.8	ゔ
L \$56- 5210.	2.660	5.046	5.812	701.6	4 + 4 CO	č.
C 556-3224.	5.627	£400.5	0,000	< 90° >	5.446	ž
156-52	2.684	165.4	0000	145.4	015.2	16
\$ 56-	•	3.050	300.0	5.630	2.120	E.
C 5 56-5244. [0.69.5	4.051	000.0	4.651	•	7.7°
C356-3254.	. 64	4.54	3.049	1.624	8.590	301
36- 526	ζ,	<. a6∪	3.	۵	6.600	101
36-36	9	5.175	100.	٧.	2.600	701

B/ST o CYNUM

TABLE 6. (Continued) Surbbakt or PHYSICAL CHARACLERIZATION HAIA

	ut ws.ly 6/cc	PLUST	PURSTY PC+	PURUS Y PC f	ANE A Mc/L	SER
556-5294.	2.635	5.612	970*0	2 1 a · c	4.030	, 6
C 556-5504.	2.700	4.434	070.0	4.459	1.050	. 16.
356-5504.2	2.711	5.66.1	070.0	4.22.4	2.478	100
136-5515.	2.113	5.076	3.745	6.869	1.670	#5 P
156-1525.	2.552	2.346	6.600	9.43p	6.860	, o .
\$56-3555.	5.614	3.400	000.0	5.400	2.500	7 1
\$56-5545.	779.2	4.210	000.0	4.670	5.400	1 1
156-1555.	2.654	7.4.5	654.	48 T. T	011.	113
556-5565.	5.613	1,510	108.	4.510	1.700	7 -
556-5575.	2.490	/10.	0,0.0	719.	. 660	
156-1584.	5.56	751.	. V34	1.685	1.100	911
156-5544.	2.546	1,624	000.0	1.224	914.	111
\$26-5404.	2,685	ns 0	000.0	. 83u	010.1	911
7.4042-955	2.705	uća.	. 455	1.584	25.7	611

	N H	STANDAND DEVÍATIDA	VARIENCE	CUEPFICIENT QF DEVIATION	ys PCT, CONF. LOVER LIME:	95 PCT, CONFIDENCE TATERVAL LOSTE LIMIT OFFER FALL	STAMVS STAMVS
BULK DENSITY, 1,7CC	450.5		10.	E 0 *	95 9.5	180.5	10.5
APPAKENT DENSITY, G/CC	2.110		20.	co.	2.684	2.150	105
INNE DENSTLY HITC	4.584	. 54	21.	\$1.	116.5	150.5	105
UPEN PURUSITY, PCI	1.050	1.41	90.5	٠٢.	1.374	4.127	1 10 5
CLUSED PURUSITY, PCT	1.00.1	C#.1	5.41	1.85	. 640	1. 500	1 3
toral posisity ref	159.2	05.5 5	4.83	11:	2.420	182.5	105
WEIGHT LUSS INTITAL, G/Kb	10.460	3.11	70.5	. 50	7.057	11.0/4	103
METGHI LUSS FINAL, G/Ku	116	6.64	44.15	45.1-	200.0	. 540	103
101AL ME19HI LUSS, GZKB	4.549	7:01	49.13	7.	0.111	10.560	105
SURFACE AREA, M276	455.5	. 4.7	4.44	٠٤.	4.646	6.031	102

AL IS INSTITUTE ICANI

24

		<u>z</u>
DENSITIES		IOTAL POROSITY, P
DATA FROM	-	IOTAL
8. POROSITY DATA FROM DENSITIES	***************************************	
TABLE 8.		SAMPLE ID.

IDIAL PONDSIIT, PCÍ.

													1							ı						1																	
																				i																							
																				į						!																	
																	- 1			ļ						į						i i											
													- 1											1		1						1											
_			S	•	•	8	•	ď	_	Э	٦.	•	اع	0	a.	'n	3	~	7	-	~	. ~	. ~			٠.	٠.			. ~	ے ا	i n	م		_	. =	m	'n	_	'n	Δ.	Λ	.
- 5	•	- 23	3	v	ď	-	0	-	>				۵										• 🔊				• 3					9		'n	~	2	~	20	J.	3	5	>	_
•	• •	•	•	•	٠	_:	•	_:	•	٠ <u>.</u>		•	-	•	•	•	•	-:	•		_		•		•	•	•	•	•	•	• •	•	•	•	•	•	٠	•	٠	•		٠	•
		!				_		_		. •		٠٠			_	•			_	۱·•	_		. –		•	i						;						:	_	_			_
													- 1							1						i						1											
											- 1		i											1		ļ												!	•				
													- 1							İ		ì				1						[
									-				i		,					ĺ		i,				İ						1											
													+											1		1						ļ		:									
		i											i							!		1										1				:							
							- 1						†											,		1		í				i				,							
							1		,				- 1							1						}						ŧ									,		
													İ							ŀ		÷				Ì						1											
													- 1							ĺ												!											į
													- 1							l		1		:																			
													- 1				- ;			1												İ											
			~				,		~ 4.				1		_							_		1			_																
•			-	•	•	٠	-	٠	-	•	•		•		-			•		١.		٠.,	: ,	,			. ``										7		-		7	٠	
7	₹.	7	4	4	4	₹	4	4	4	4	₹.	Š	Š	Š	S.	S	•	2	Š	-	-	. ~	. >	ه	9	1	4	: 4	4), ¢	S	Ŋ	S	S	M	M	₹	4	٦.	Č	N.	٠,	-
3	4	3.	2	=	3	3	3	3	3	Š	2	4	٠, در	Š	5	5	3	.2	3		5			3		1			•	: 3	=	2	3	Ţ	٥	~	3	=	3	7	3	T.	_
Ñ	1 W	•	~	N	\sim	∿	~ !	ಌ	~	Ą	٠,	v	71	U	v	าง	7	~s	N	17	~	· ~	ı N		T T	HA.	1 3	<u>ا</u> م	١٦	1 1	1 ~	· ~	∼પ	1	~	· ~	~	N	₹	ž	ž	٦̈	ũ
1									•	ŗ	Į.	Ŀ	1	ŗ	•	1	Ţ	1	ŀ	<u> </u>	•	. 1		١	•	11			ŧ			Ħ	ŧ	t	1	ŧ	ī	1	1	ŧ	Ĺ	•	ı
~	5	×	×	š	×	×	×.	×	¥.	×	3	ž	*	×	3	ž	3	3	20	3	36	3.6	36		3	1 5	5		9	3	36	. 5	36	. ≅	30	56	Š	3	\$0	\$6	\$	\$	~
~	•	•	•	~	~	~	~ 1	**	∽,	~	•	9	*	~	₹,	~	*	. 77	**	•	•	. •	*	. 7	. •	11.7		. •	•	j: 🕶	-	. •	~	. 🕶	~	•	*	*	٧٩.	.~1		٠,	4
J	د ر	ں	٠.	ပ	U	u	ان	ပ	ပ	J	ن.	Ç	ن	Ç	Ç	Ç	J	ပ	ပ	10	J	ں .	ני	ני	·		, –		9	ب :	U	Ü	ပ	Ų	C	Ů	J	C	Ç	Ç	Ľ	Ü	J
													- 1							į.																							
													1													1																	
							,						4							1						1																	
							i						1							1						1																	
													- 1							i						ŧ																	
																										i -																	

69.	5.5				190	¥.	2 4	. ₹.	. i.	15.	. 56		66.	ه. د	9 .	5	.	20.0	, ,		60	7	4: :	•	3 30	-		3	٧.	3	9	3 0::	Λ.	•	•	٠.	9:	2	₹9.
-	-		-		1		Λ.	•	~		-	٠.				Ì				~1	~	*	3. 4	^ ~	٠~٦	. ~	· 3	~ 3·	7	•	₹7	~	7	~ }1	7	37 1 €	u ,	•	7
					į									:				1		:							ŀ					1							
	i				i							:				İ				:							1	,				!		:				i	
	:				ļ									:		1				:												1		1				1	
		:										:		1									•				ķ	:		ı				,					
												,		1				-										:		:		i							
	1																				ļ		:				1					1						i	
•	7:	•		•	•	7.	•1	• •		•	~	•		•									7									•		•	7.	•	•		
		 ~	. ~	0 16	91.	999		, ,	. 🌫	-	-	v	*	3"	so.	۰۱۹	: ٥	20 ⋅ 0	•	> ~	127	.~	~ ••••	*	7 4	~	- 20	ு	3	3	-	~;;	•	₹.	₹.	ე, .	ø,	7	
, ,		1		1	٠.	t I		t	7	~	~	~	~	٠,١		71:	j	ا دارد	? ~	1 1	5-9	5	7 · ·	? 1) N		•	2.	7	7	?	4	٤.		۵.		į.	٠٠. ا	Š
-	5	-		١ (١٠)	اذات	3		1	3	4	, Y	~	*	٠,	χή ;	٠. د اور	3	ماري ما . لوه	7	Ä	P 7:	٠ ٠ ٠	M) N	'n	يما : ل	-		1	~	~	~		~	٠,	•	3	٠, i		~
<u>ي</u> د	ن اد	ی ن	, 0	J	ات ا	ن ر	ے . ر	ט נ	ن .	J	U	٠.	J	۱	ان	١	، د	ن∷د	ے د	ں د	3	ن	ن ت	ے د	ے د	ניו	ت	J	J	J	u	۰۰۰	ن ن	.	٠	ا ب	، ت	۰. ر	ن

•	•	चर	'n
	3.	7	•
•	•	-	~
	•	-	
•			
(
3			
-		į	
1.			
		- 1	
		- 4	
-			
3			
<u>.</u>			
-			
		i	
		4	
		*	
		- 1	
		- !	
		-	
			~
		•	

				1			
				!			
	!						
	1						
							- 1
	1						
	•			1			
				- 1			
	!						
	:						
				a de l			_
	~						~
• •							
* *					• •		3
		٠. ٠. ک		Sign	₹.	4 4	
~ ~	Ċ	4.5°	** 3	515	15 B 4.	3 T	9
2.5	2	315. 325.	35.	365	375.	544°	3
329	330	3315°	333	3365	3375. 5384.	4344	346
-329	-330	-3315. -3325.	-333	- 4355 - 3365	-3375.	-3404	-346
-329	-330	-3315. -3325.	-333	- 4355 - 3365	-3375.	-3404	-346
-329	-330	-3315. -3325.	-333	- 4355 - 3365	-3375.	-3404	-346
36-329	36-330	30-3315.	36-334	36-3355	36-3375.	30-3594	36-346
530-329 556-550	136-330	536-3315. 536-5325.	456-434	436-3355	\$30-\$375. \$56-\$384.	330-3394	336-346
530-329 556-550	136-330	536-3315. 536-5325.	456-434	36-3355	\$30-\$375. \$56-\$384.	330-3394	336-346
530-329 556-550	136-330	536-3315. 536-5325.	456-434	436-3355	\$30-\$375. \$56-\$384.	330-3394	336-346
530-329 556-550	136-330	536-3315. 536-5325.	456-434	436-3355	\$30-\$375. \$56-\$384.	330-3394	336-346

TABLE 9. MERCURY INTRUSION POROSITIES FOR WELL C-336

SEQUENCE	SAMPLE	POROSITY,
NUMBER	ID	%
1	2434	4.10
2	2444	6.40
3	2454	3.71
4	2454.Z	2.12
5	2463	4.24
6	2474	3.44
7	2494	4.24
8	2504	3.71
9	2514	6.10
10	2514.Z	4.50
11	2524	4.10
12	2534	4.10
13	2545	8.10
14	2555	4.24
15 16	2565	5.30
16 17	2565.Z	5.30
18	2575 2586	4.50
19	2586	9.54
20	2588	5.10
21	2595 2607	3.71
22	2617	4.10 5.10
23	2617.Z	
24	2637	3.44 3.44
25	2646	5.03
26	2656	5.30
27	2666	5.03
28	2666.Z	1.10
29	2676	3.01
30	2686.A	4.10
31	2686.B	2.65
32	2696	5.10
33	2705	5.10
34	2715	5.03
35	2725	10.10
36	2725.Z	6.10
37	2735	5.10
38	2745	3.01
39	2755	4.10
40	2763	5.03

TABLE 9. (Continued)

43	2784.Z	3.20	
44	2793	4.10	
45	2802	8.21	
46	2814	4.10	
47	2822	2.65	
48	2830	9.01	
49	2842	4.10	
50	2842.Z	5.03	
51	2852	3.44	
52	2862	6.41	
53	2872	5.61	
54	2881	9.80	
55	2891	4.50	
56	2901	5.03	
57	2901.Z	5.10	
58	2911	4.50	
59	2921	4.10	
60	2931	4.24	
61	2940	5.30	
62	2950	4.10	
64	2960.Z	3.20	
65	2970	4.24	
66	2980	5.83	
67	2990	5.03	
68	2999	3.44	
69	3009	4.10	
70	3019	3.71	
71	3019.Z	4.50	
72	3029	6.62	
73	3038	4.50	
74	3049	7.71	
75	3054	4.24	
76	3069	3.71	
77	3069.Z	3.20	
78	3079	5.10	
79	3089	4.10	
80	3099	5.10	
81	3107	4.10	
82	3117	4.10	
83	3127	4.10	
84	3137	7.42	
9 * r	J 1 7 1	/ • 7 <u>~</u>	

TABLE 9. (Continued)

85	3137.Z	9.80	
86	3137.2	7.10	
87	3157	6.41	
88	3166	7.15	
89	3176	10.91	
90	3186	5.10	
91	3186.Z	3.71	
92	3196.A	5.30	
93	3196.B	6.41	
94	3206	5.10	
95	3216	3.20	
96	3224	7.10	
97	3234	5.61	
98	3244	5.61	
99	3244.Z	7.15	
100	3254	5.10	
101	3264	3.01	
102	3274	6.41	
103	3284	6.62	
104	3294	9.80 *	
105	3304	5.83	
106	3304.Z	9.01	
108	3315	5.83	
109	3325	5.10	
110	3335	5.61	
111	3345	5.31	
112	3345.Z	7.42	
113	3355	5.03	
114	3365	4.24	
115	3375	4.24	
116	3384	2.65	
117	3394	8.10	
118	3404	3.01	
119	3404.Z	7.15	

TABLE 10. ENERGY DISPERSIVE ANALYSIS OF SHALES

					H	Element Count Per 100 Counts	Count	er 100	Count	S				
Shale Sample	Na	X	Са	Mg	A1	Si	Fe	S	Ti	Ь	Fe/S	K/A1	K/Si	A1/S1
C-336-2524	1	12.4	l	2.0	21.0	52.0	10.7	t	1.9	1	8	0.59	0.24	0.40
C-336-2646	í	8.1	t	2.0	26.4	59.7	3.3	!	0.5	ı	8	0.31	0.14	0.44
C-336-2763	į	7.7	2.8	Tr	19.9	8.09	5.0	2.8	1.0	ŧ	1.8	0.39	0.13	0.33
C-336-2842ZC	i	10.8	ı	1.8	24.4	56.3	4.6	6.0	1.2	1	5.0	0.44	0.19	0.43
C-336-2990	1	5.5	i	Tr	17.3	70.1	3.3	3.4	9.0	1	1.0	0.32	0.08	0.25
C-336-3186ZCB	ı	7.1	7.4	Tr	20.4	51.7	6.1	6.3	1.1	4	1.0	0.35	0.14	0.39
C-336-3315	ł	10.1	ı	Tr	22.2	61.4	4.5	6.0	6.0	1	5.3	0.46	0.16	0.36

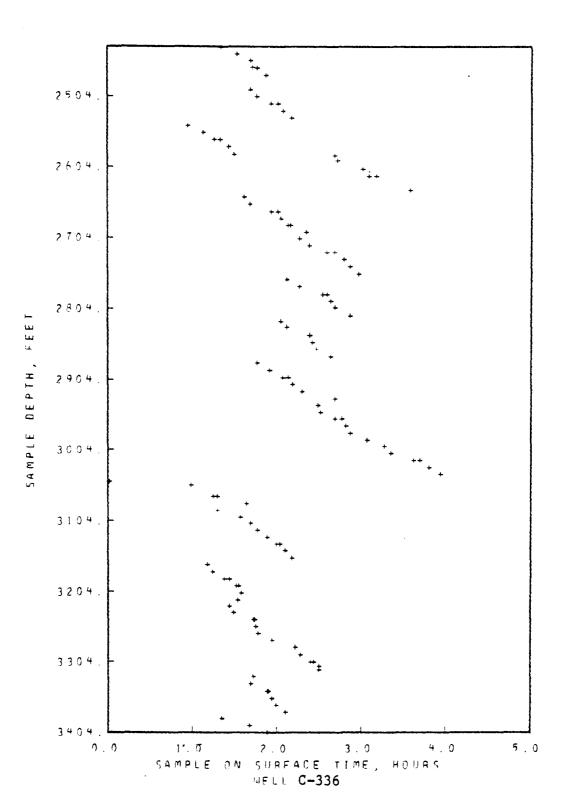


FIGURE 1. ON-SURFACE TIME FOR WELL C-336

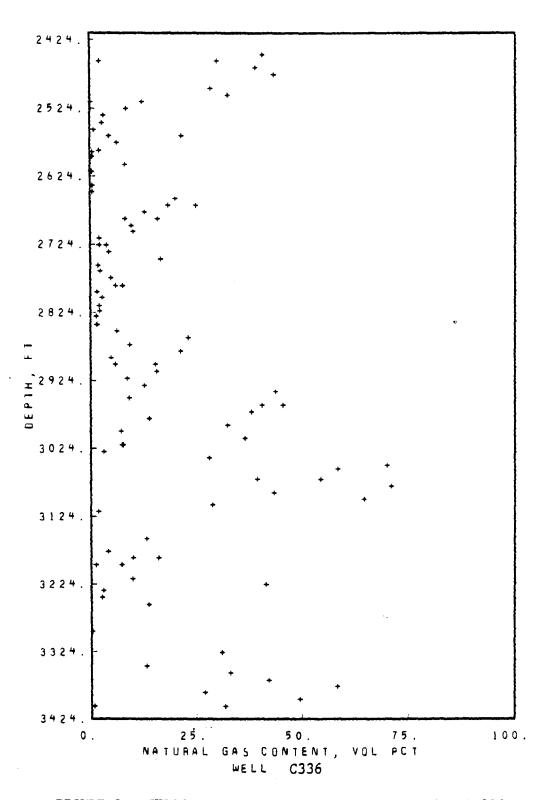


FIGURE 2. HYDROCARBON GAS VERSUS DEPTH FOR WELL C-336

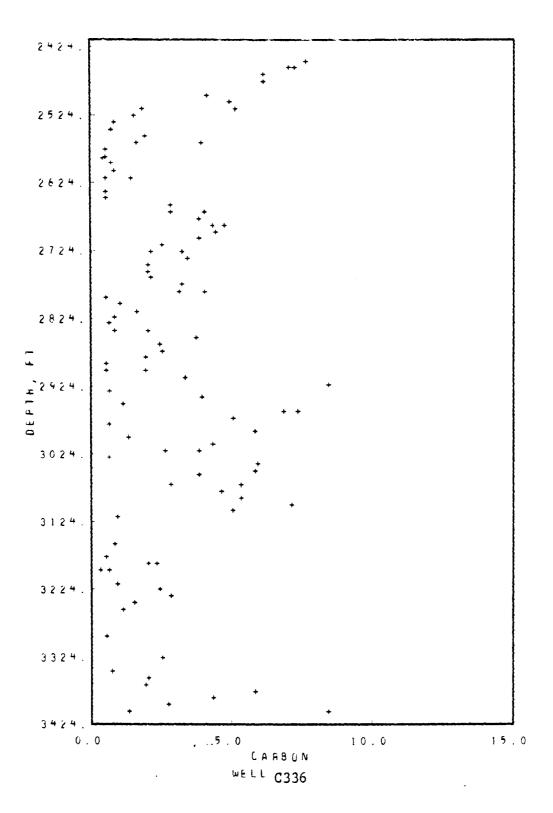


FIGURE 3. CARBON VERSUS DEPTH FOR WELL C-336

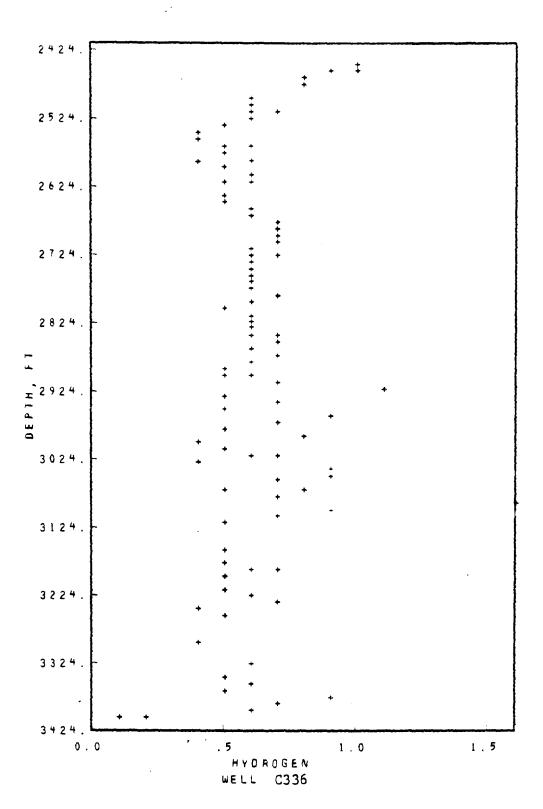


FIGURE 4. HYDROGEN VERSUS DEPTH FOR WELL C-336

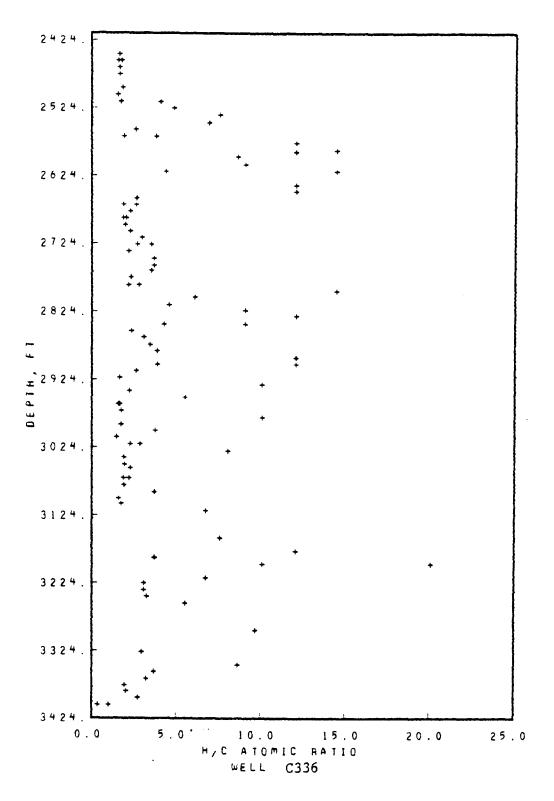


FIGURE 5. H/C ATOMIC RATIO VERSUS DEPTH FOR WELL C-336

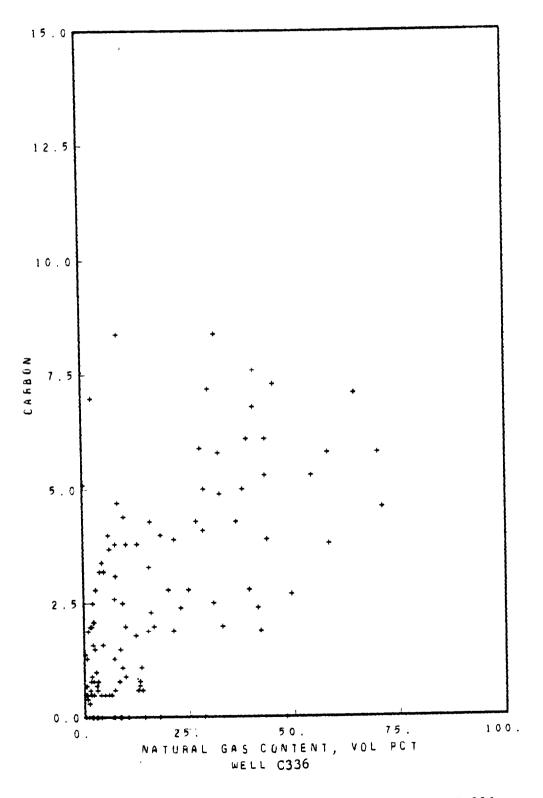


FIGURE 6. NATURAL GAS VERSUS CARBON FOR WELL C-336

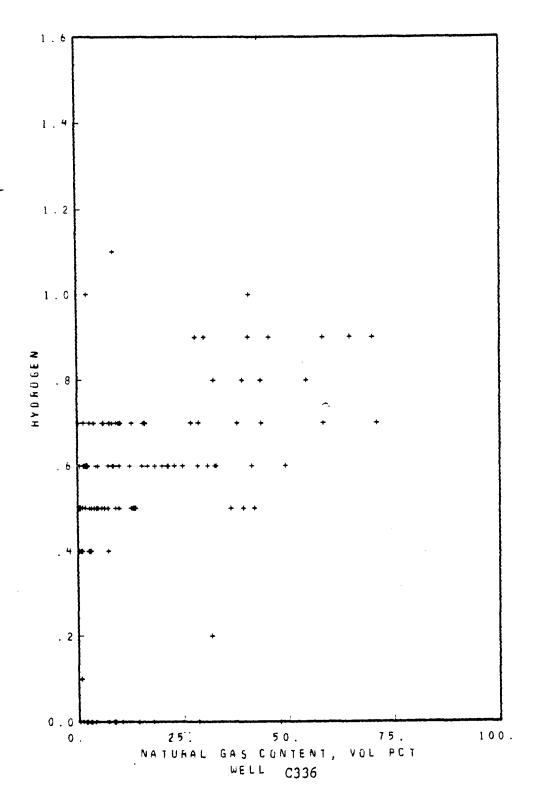


FIGURE 7. NATURAL GAS VERSUS HYDROGEN FOR WELL C-336

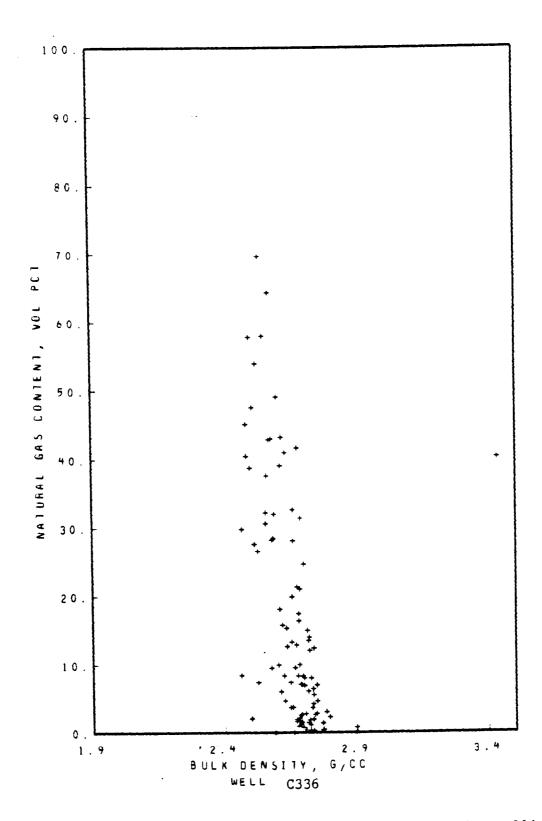


FIGURE 8. HYDROCARBON GAS VERSUS BULK DENSITY FOR WELL C-336

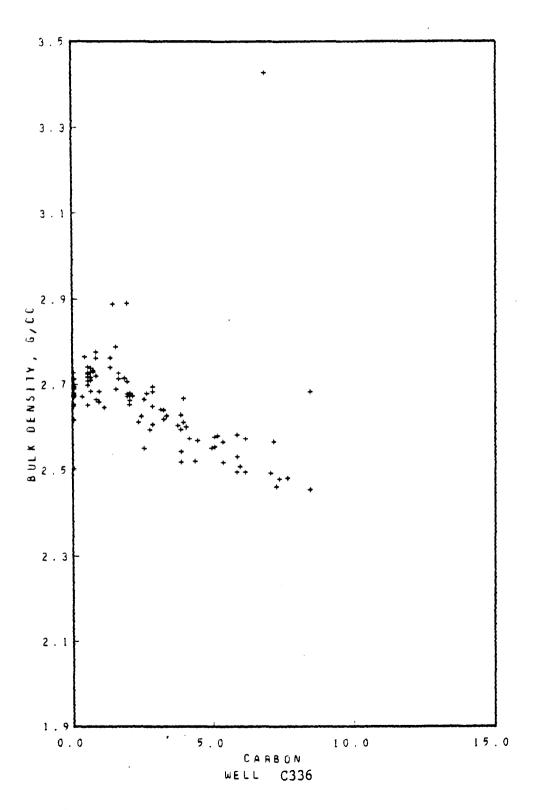


FIGURE 9. CARBON VERSUS BULK DENSITY FOR WELL C-336

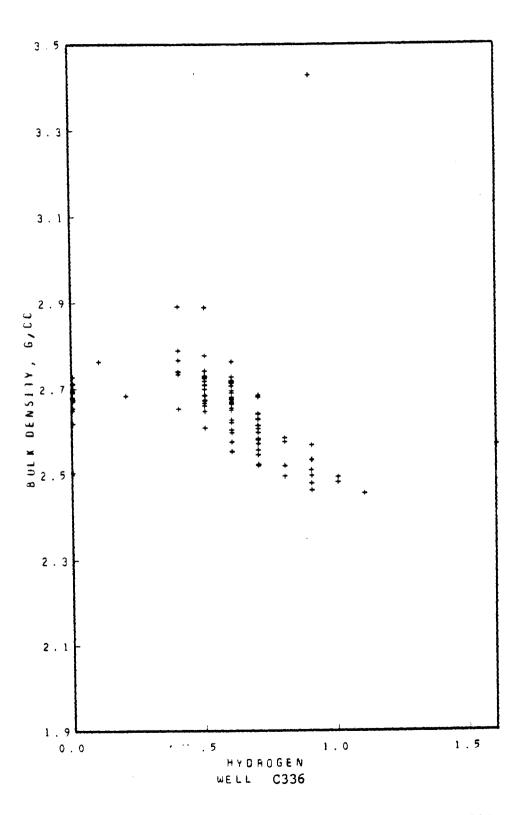


FIGURE 10. HYDROGEN VERSUS BULK DENSITY FOR WELL C-336

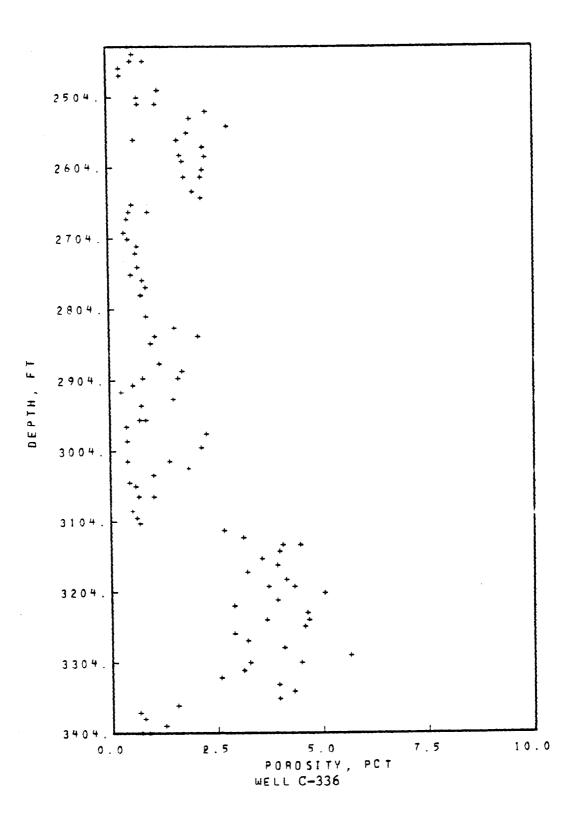


FIGURE 11. POROSITY VERSUS DEPTH FOR WELL C-336

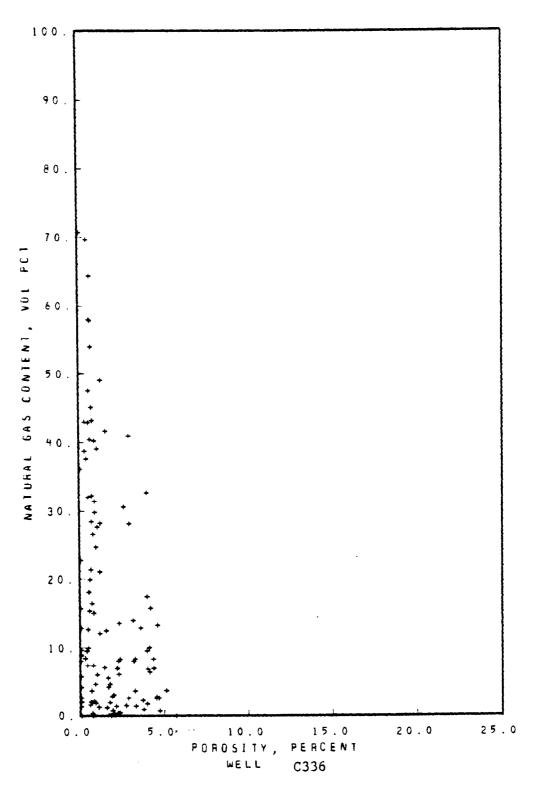


FIGURE 12. POROSITY VERSUS NATURAL GAS FOR WELL C-336

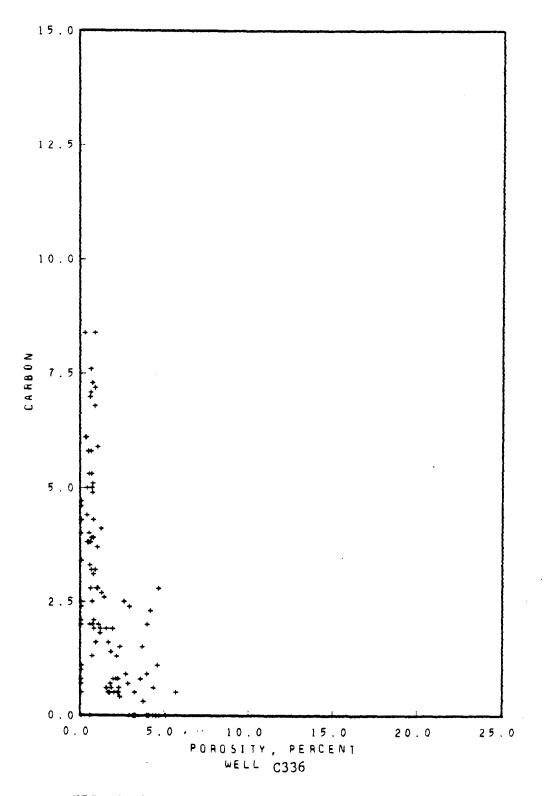


FIGURE 13. POROSITY VERSUS CARBON FOR WELL C-336

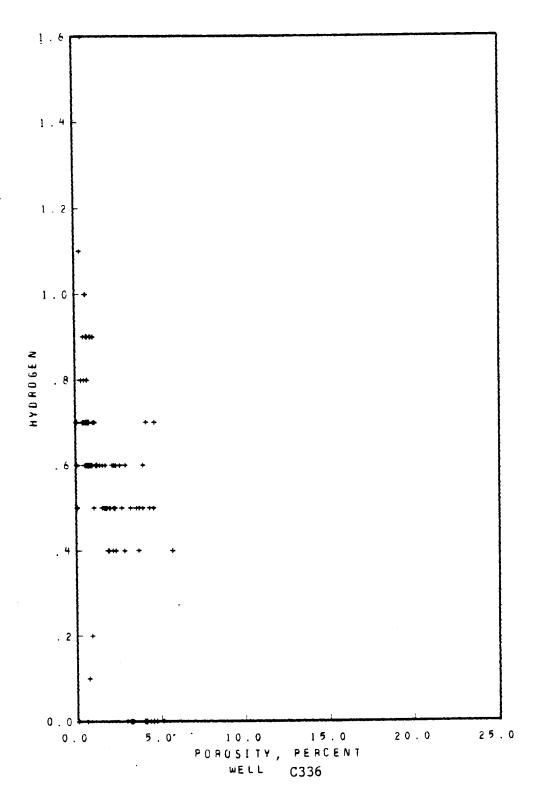
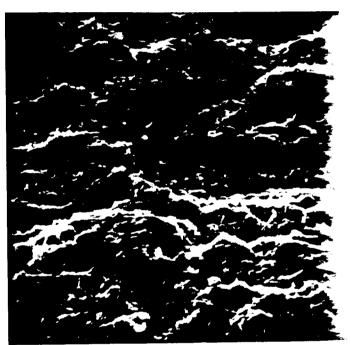


FIGURE 14. POROSITY VERSUS HYDROGEN FOR WELL C-336



700X

FIGURE 15. SEM MICROGRAPH OF SAMPLE C-336-3842



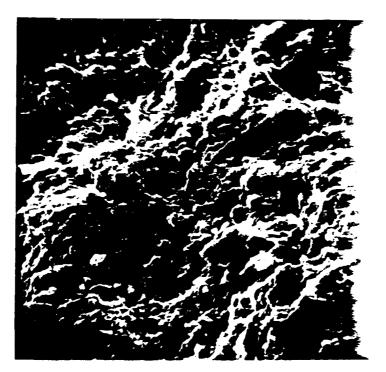
700 X

FIGURE 16.
SEM MICROGRAPH OF SAMPLE C-336-3315



700 X

FIGURE 17. SEM MICROGRAPH OF SAMPLE C-336-3186



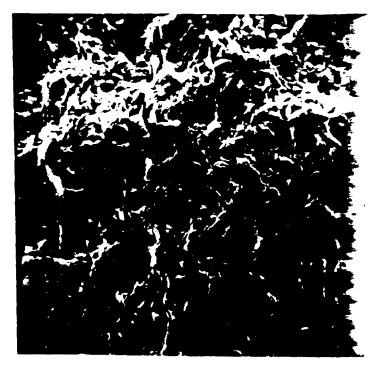
700 X

FIGURE 18.
SEM MICROGRAPH OF SAMPLE C-336-2990



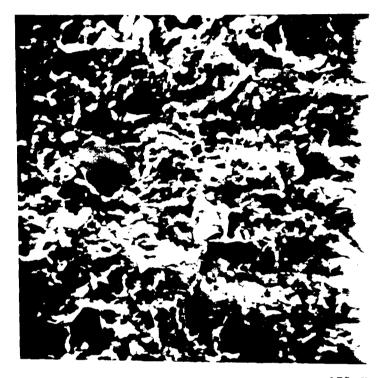
700 X

FIGURE 15. SEM MICROGRAPH OF SAMPLE C-336-2763



700 X

FIGURE 20. SEM MICROGRAPH OF SAMPLE C-336-2646



675 X

FIGURE 21. SEM MICROGRAPH OF SAMPLE C-336-2524

د د

C-336 HYDROCARBON GAS VERSUS LABORATORY AND WELL LOG DATA FIGURE 22.

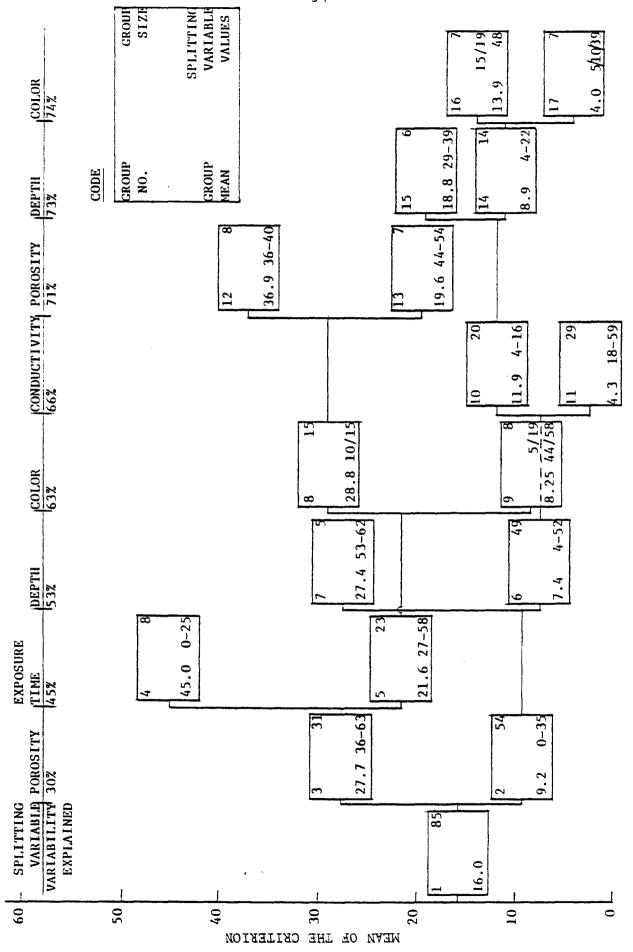
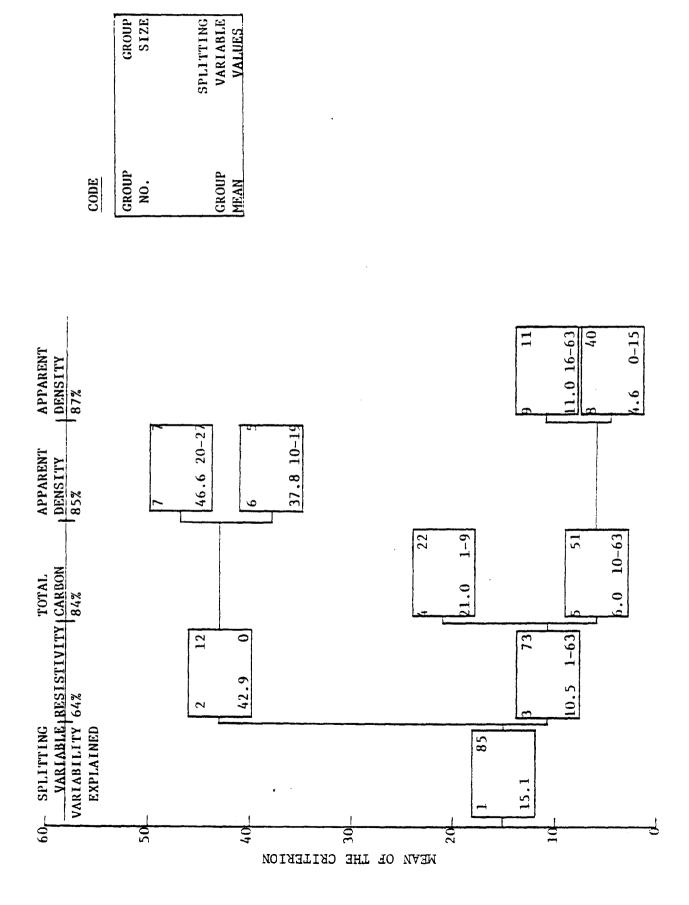


FIGURE 23. C-336 HYDROCARBON GAS VERSUS WELL LOG DATA



C-336 OPEN POROSITY VERSUS LABORATORY AND WELL LOG DATA FIGURE 24.

